



Emission spectra of the brass sample were detected and recorded under time-integrated mode of OMA.

### Results and Discussion

Fig.2 shows how temporally and spatially integrated emission intensities of Zn and Cu vary with the number of repeated laser irradiation on the standard brass sample (NBS C 1118, 75 % Cu, 25 % Zn) at a fixed position. In case of TEA CO<sub>2</sub> laser, the emission intensity of Cu I 515.3 nm spectral line could be detected only during the first and second shot with much lower intensity, despite the fact that the concentration of Cu was actually higher than that of Zn in the target.

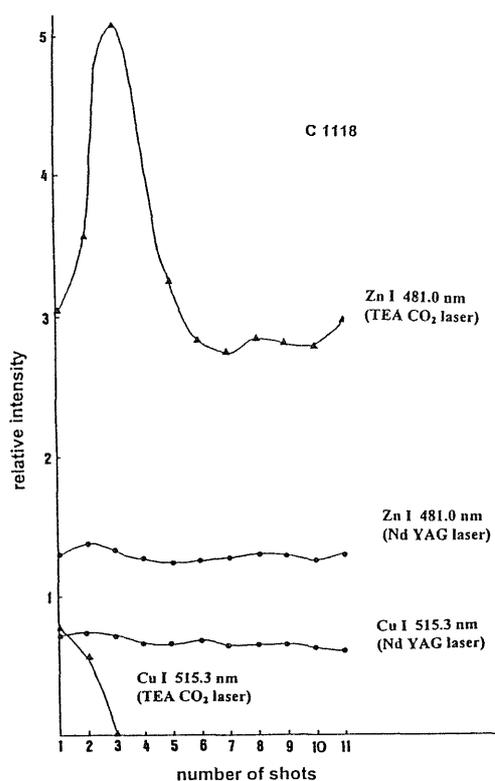


Fig. 2. Relationship between total emission intensity (time and spatially integrated) of Zn I 481.0 nm and Cu I 515.3 nm as a function of the number of shots of laser irradiation for TEA CO<sub>2</sub> laser and Nd-YAG laser

A different result was obtained when Nd-YAG laser was used, as shown by the corresponding curves in Fig.2. Emission intensities of both Zn and Cu spectral lines remained nearly constant under the repeated irradiation at the same position. The power density for Nd-YAG laser employed was 5 times stronger (20 GW/cm<sup>2</sup>) than the case of TEA CO<sub>2</sub> laser due to the much higher beam quality of the Nd-YAG laser

It is supposed that atoms of elements with lower boiling point are evaporated sooner and gushed out from the target surface with high velocities to form a shock wave in the surrounding gas. Namely only the Zn atoms contributed to the formation of shock wave front, whereas the Cu atoms were left behind the shock wave and remained unexcited. The selective vaporization appeared to be less pronounced at the laser power density considerably higher than threshold of plasma generation. This is the case for Nd-YAG laser in Fig. 2.

In order to see more clearly the evaporation process, a comparison of time profiles for Zn I 481.0 nm and Cu I 327.4 nm emission intensities was conducted using a dual channel measurement of the emission at 5 mm from the surface. The result is shown in Fig. 3. It is perceptibly clear that the emission of Cu took place later and continued for long time. The Zn emission, on the other hand, started earlier and lasted only a short time.

It is also experimentally proved that selective vaporization in favour of Zn atoms takes place even for Nd-YAG laser irradiation to a certain extent. The effect of surrounding air pressure and repeated irradiation on the selective vaporization for Nd-YAG laser was studied in detail using OMA system, the result will be also reported.

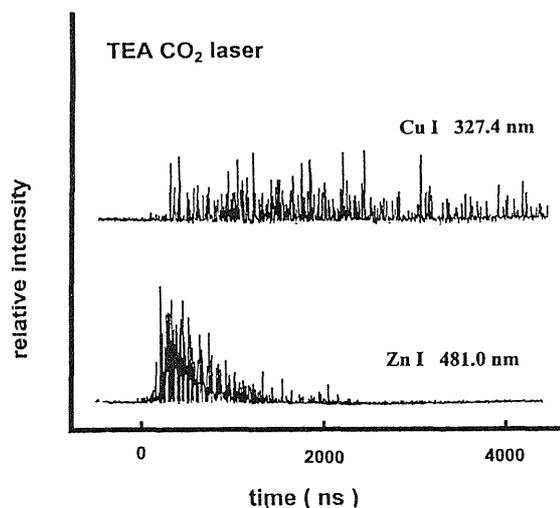


Fig. 3. Time-profile of the emission intensity of Cu I 327.4 nm and Zn I 481.0 nm using TEA CO<sub>2</sub> laser. Data was collected at 5 mm above the sample surface at a reduced pressure of 1 Torr.

### References

1. H. Kurniawan, M. Pardede, K. Kagawa, and M.O. Tjia: J. Spectrosc. Soc. Japan, 47, 5 (1998)
2. R.R Russo: Appl. Spectrosc. 49, 14A (1995)